metal-organic papers

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Key indicators

Single-crystal X-ray study T = 293 K Mean $\sigma(C-C) = 0.009 \text{ Å}$ R factor = 0.042 wR factor = 0.117 Data-to-parameter ratio = 15.0

For details of how these key indicators were automatically derived from the article, see http://journals.iucr.org/e.

catena-Poly[[silver(I)-µ-1,4-diaminobutane- $\kappa^2 N:N'$] hexafluoroarsenate]

The title compound, $\{[Ag(C_4H_{12}N_2)](AsF_6)\}_n$, is a helical onedimensional chain Ag^I complex. The Ag^I atom is coordinated by two N atoms from two 1,4-diaminobutane ligands, in a nearly linear geometry. The crystal structure is a threedimensional structure stabilized by N-H···F hydrogen bonds and weak Ag · · · F interactions.

Comment

The title complex, (I), crystallizes in the monoclinic space group $P2_1/n$. The complex consists of polymeric 1,4-diaminobutanesilver(I) cations and hexafluoroarsenate counterions as shown in Fig. 1. The simplest repeat unit is the 1,4-diaminobutanesilver(I) cation and a hexfluoroarsenate anion. In the cation, the Ag atom is in a linear coordination environment, being coordinated by two N atoms from different butanediamine ligands. The average Ag-N distance is 2.137 (5) Å and the angle around Ag1 is $178.1 (2)^{\circ}$, indicating a slightly distorted linear geometry at the Ag1 atom.



All the butane ligands are fully extended in different directions and are linked together by the Ag-N bonds, forming a one-dimensional chain along the *c* axis (see Fig. 2).

In the crystal structure, extensive intermolecular hydrogen bonds $(N2-H2C\cdots F1, N1-H1C\cdots F1^{i})$ and $N2-H2D\cdots F2^{i}$; for symmetry codes see Table 1), and weak interactions between the Ag atom and the F atoms $[Ag1 \cdot \cdot \cdot F3 =$ 3.462 (7) Å, Ag $1 \cdot \cdot \cdot F4^{iii}$ = 3.353 (7) Å, Ag1...F5^{iv} =



Figure 1

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The structure of the title compound, showing 20% probability displacement ellipsoids and the atom-numbering scheme.

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3.241 (7) Å, Ag1...F6 = 3.208 (7) Å and Ag1...F6^{iv} = 3.126 (7) Å] were observed [symmetry codes: (iii) $x - \frac{1}{2}$, $-y + \frac{3}{2}$, $z + \frac{1}{2}$; (iv) -x, -y + 2, -z + 1].

Experimental

All reagents and solvents were used as obtained without further purification. AgAsF₆ (0.5 mmol, 148 mg) and 1,4-diaminobutane (0.5 mmol, 44 mg) were dissolved in ammonia (10 ml). The mixture was stirred for *ca* 10 min to give a clear solution. After allowing the solution to stand in air for 3 d with gradual loss of ammonia gas, large colorless crystals were formed. The crystals were filtered off and washed three times with water and dried in a vacuum desiccator using anhydrous CaCl₂ (yield 63%). Elemental analysis found: C 12.59, H 3.21, N 7.18%; calculated for C₄H₁₂AgAsF₆N₂: C 12.48, H 3.14, N 7.28%.

 $D_x = 2.480 \text{ Mg m}^{-3}$

Cell parameters from 4450

Mo $K\alpha$ radiation

reflections

 $\theta = 2.8 - 25.4^{\circ}$ $\mu = 5.19 \text{ mm}^{-1}$

T = 293 (2) K

Block, colorless

 $0.38 \times 0.32 \times 0.20$ mm

Crystal data

$$\begin{split} & [\mathrm{Ag}(\mathrm{C_4H_{12}N_2})](\mathrm{AsF_6}) \\ & M_r = 384.95 \\ & \mathrm{Monoclinic}, P2_1/n \\ & a = 7.806 \ (2) \ \mathrm{\AA} \\ & b = 10.371 \ (2) \ \mathrm{\AA} \\ & c = 13.234 \ (3) \ \mathrm{\AA} \\ & \beta = 105.74 \ (3)^{\circ} \\ & V = 1031.2 \ (4) \ \mathrm{\AA}^3 \\ & Z = 4 \end{split}$$

Data collection

Siemens SMART CCD area-	2025 independent reflections
detector diffractometer	1757 reflections with $I > 2\sigma(I)$
φ and ω scans	$R_{\rm int} = 0.030$
Absorption correction: multi-scan	$\theta_{\rm max} = 26.0^{\circ}$
(SADABS; Sheldrick, 1996)	$h = -9 \rightarrow 9$
$T_{\min} = 0.155, T_{\max} = 0.366$	$k = -12 \rightarrow 8$
4479 measured reflections	$l = -14 \rightarrow 16$

Refinement

```
      Refinement on F^2
      w = 1/[\sigma^2(F_o^2) + (0.0714P)^2

      R[F^2 > 2\sigma(F^2)] = 0.042
      w = 1/[\sigma^2(F_o^2) + (0.0714P)^2

      wR(F^2) = 0.117
      where P = (F_o^2 + 2F_c^2)/3

      S = 1.08
      (\Delta/\sigma)_{max} = 0.001

      2025 reflections
      \Delta\rho_{max} = 1.22 \text{ e } \text{ Å}^{-3}

      135 parameters
      \Delta\rho_{min} = -0.90 \text{ e } \text{ Å}^{-3}

      H atoms treated by a mixture of independent and constrained refinement
      \sigma_{max} = 0.001 \text{ e } \text{ Å}^{-3}
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Table 1

Hydrogen-bonding geometry (Å, °).

$D - H \cdots A$	D-H	$H \cdot \cdot \cdot A$	$D \cdots A$	$D - H \cdots A$
$N2-H2C\cdots F1$ $N1-H1C\cdots F1^{i}$ $N2-H2D\cdots F2^{ii}$	0.91 (8)	2.17 (8)	2.987 (8)	150 (7)
	0.90	2.11	3.012 (7)	175
	0.81 (7)	2.23 (7)	3.018 (8)	165 (6)

Symmetry codes: (i) $x - \frac{1}{2}, \frac{3}{2} - y, \frac{1}{2} + z$; (ii) $x - \frac{1}{2}, \frac{3}{2} - y, z - \frac{1}{2}$.



Figure 2

The one-dimensional helical chain of (I), viewed along along c.



Figure 3 The crystal packing of (I), showing the hydrogen-bonding interactions as dashed lines.

H atoms H2C and H2D were located in Fourier maps and were refined isotropically. All the other H atoms were placed in geometrically idealized positions and constrained to ride on their parent atoms, with N-H and C-H distances of 0.90 and 0.96 Å, respectively; the U_{eq} values for these H atoms were fixed at 0.08 Å². The F atoms have quite large U_{eq} values, but they were not considered disordered. The highest peak is located at (0.2765, 0.2301, 0.0467).

Data collection: *SMART* (Siemens, 1996); cell refinement: *SMART*; data reduction: *SAINT* (Siemens, 1996); program(s) used to solve structure: *SHELXS*97 (Sheldrick, 1997*a*); program(s) used to refine structure: *SHELXL*97 (Sheldrick, 1997*a*); molecular graphics: *SHELXTL* (Sheldrick, 1997*b*); software used to prepare material for publication: *SHELXTL*.

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